

4-(Trifluoromethyl)quinoline Derivatives

Olivier Lefebvre,^[a] Marc Marull^[a] and Manfred Schlosser*^[a,b]

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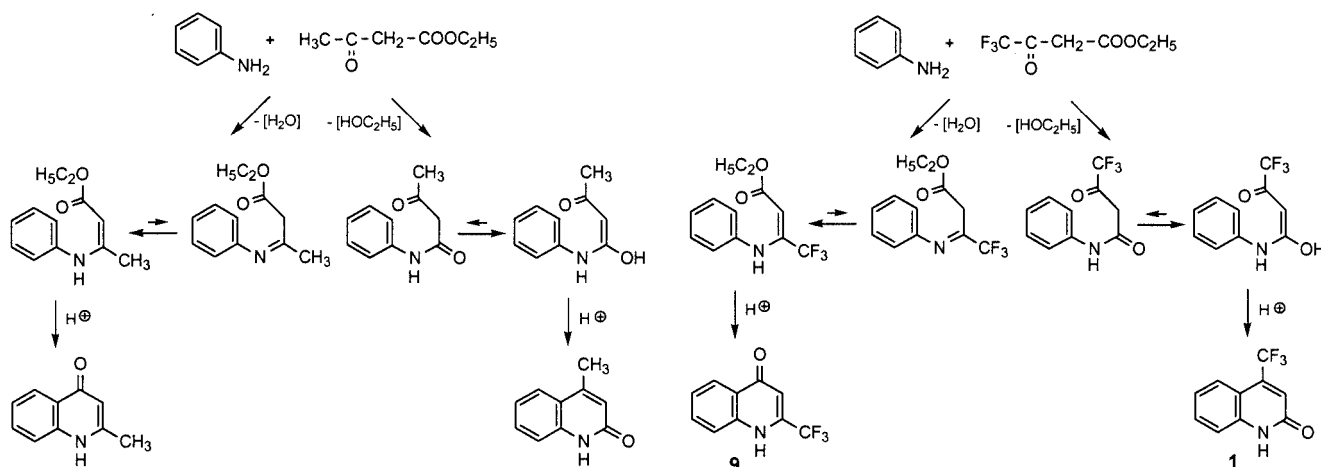
Under carefully controlled conditions, ethyl 4,4,4-trifluoroacetoacetate (ethyl 4,4,4-trifluoro-3-oxobutanoate) can be condensed with anilines and subsequently cyclized to give 4-trifluoromethyl-2-quinolinones **1** although only in poor yield. Heating these products with phosphoryl tribromide affords 2-bromo-4-(trifluoromethyl)quinolines **2** which can be converted into 4-(trifluoromethyl)quinolines **3** by reduction, 4-trifluoromethyl-2-quinolinecarboxylic acids **4** by permutational halogen/metal exchange followed by carboxylation,

and 2-bromo-4-trifluoromethyl-3-quinolinecarboxylic acids **5** by consecutive treatment with lithium diisopropylamide and dry ice. Debromination of acids **5** makes 4-trifluoromethyl-3-quinolinecarboxylic acids **6** available. As at any time 2-trifluoromethyl-4-quinolinones **9** may form instead of the expected isomers **1**, the structures have to be assigned on the basis of unequivocal NMR spectral criteria.

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“To kill two birds with one stone” is a saying that finds its chemical equivalent in the conversion of the same starting material into two different products. An early example was published more than a century ago by R. Knorr.^[1] Aniline and ethyl acetoacetate react with each other at ambient temperature to form an imino ester (in equilibrium with the tautomerically predominant ethyl 3-anilino-2-butenate) which gives 4-quinolinone upon acid-catalyzed cyclization^[1] whereas the acetoacetanilide (or the tautomeric enol) and, after acid treatment, the 2-quinolinone are obtained when the condensation between aniline and the dicarbonyl compound is carried out at temperatures around 150 °C.^[1,2]

The same dichotomy is encountered with ethyl 4,4,4-trifluoroacetoacetate, a versatile technical bulkware.^[3] When combined with aniline in the presence of polyphosphoric acid at 150 °C, it produces the anil, which predominantly exists in the tautomeric 3-amino-2-butenate form. Ring closure to the 2-trifluoromethyl-4-quinolinone occurs upon prolonged heating to 170 °C.^[4] However, if aniline and ethyl trifluoroacetate are very briefly kept at 235 °C, the 4,4,4-trifluoroacetoacetanilide (or the enol-like tautomer) is obtained. Subsequent treatment with concentrated sulfuric acid at 95 °C yields the 4-trifluoromethyl-2-quinolinone.^[5–7]



^[a] Institut de Chimie moléculaire et biologique Ecole Polytechnique Fédérale, BCh 1015 Lausanne, Switzerland

^[b] Faculté des Sciences, Université, BCh 1015 Lausanne, Switzerland
E-mail: manfred.schlosser@epfl.ch

As described in a preceding article,^[3] we have prepared a considerable number of 2-trifluoromethyl-4-quinolinones and have converted them into several families of functionalized, mainly carboxy-substituted, derivatives. We have

The reaction scheme illustrates the synthesis of 2-bromo-3-(trifluoromethyl)-6-R-quinoline-4-carboxylic acid (5) from 2-amino-6-R-quinoline-4-carboxylic acid (2) and ethyl 3-oxo-3-(trifluoromethyl)butyrate. The scheme shows two parallel pathways from the starting materials to intermediate 3, which then leads to the final product 5.

Starting Materials:

- 2-amino-6-R-quinoline-4-carboxylic acid (2): A quinoline ring with a trifluoromethyl group (CF_3) at position 3, a bromine atom (Br) at position 2, and an R group at position 6.
- Ethyl 3-oxo-3-(trifluoromethyl)butyrate: $\text{F}_3\text{C}-\text{C}(\text{O})-\text{CH}_2-\text{CO}_2\text{C}_2\text{H}_5$.

Reaction Pathways:

- Pathway 1 (Top):** 2-amino-6-R-quinoline-4-carboxylic acid (2) reacts to form 2-bromo-3-(trifluoromethyl)-6-R-quinoline-4-carboxylic acid (5).
- Pathway 2 (Bottom):** 2-amino-6-R-quinoline-4-carboxylic acid (2) reacts to form 2-bromo-3-(trifluoromethyl)-6-R-quinoline-4-carboxylic acid (5).

Intermediate 3: 2-bromo-3-(trifluoromethyl)-6-R-quinoline-4-carboxylic acid (5) is the common intermediate for both pathways.

Final Product: 2-bromo-3-(trifluoromethyl)-6-R-quinoline-4-carboxylic acid (5).

Table 1. 4-Trifluoromethyl-2-quinolinones **1** by condensation of ethyl 4,4,4-trifluoroacetate with a variety of anilines followed by the acid-catalyzed cyclization of the amide intermediate

Compound	Precursor	R	Yield ^[a]
1a	aniline	H	34% ^[b]
1b	<i>p</i> -toluidine	6-H ₃ C	32%
1c	<i>p</i> -fluoroaniline	6-F	11%
1d	<i>m</i> -toluidine	7-H ₃ C	29%
1e	<i>m</i> -anisidine	7-H ₃ CO	30%
1f	<i>m</i> -fluoroaniline	7-F	31%
1g	<i>o</i> -toluidine	8-H ₃ C	21%

The conversion of the quinolinones **1** into the 2-bromoquinolines **2** was accomplished with phosphoryl tribromide by applying a standard procedure. The yields were acceptable, though far from quantitative (see Table 2).

Compound	R	Precursor	Yield
2a	H	1a	66%
2b	6-H ₃ C	1b	59%
2c	6-F	1c	59%
2d	7-H ₃ C	1d	62%
2e	7-H ₃ CO	1e	59%
2f	7-F	1f	51%
2g	8-H ₃ C	1g	64%

Compound	R	Precursor	Yield
3a	H	2a	63%
3b	6-H ₃ C	2b	72%
3c	6-F	2c	68%
3d	7-H ₃ C	2d	71%
3e	7-H ₃ CO	2e	68%
3f	7-F	2f	55%
3g	8-H ₃ C	2g	74%

Table 4. 4-Trifluoromethyl-2-quinolinecarboxylic acids **4** from 2-bromoquinolines **2** by consecutive reaction with butyllithium and dry ice

Compound	R	Precursor	Yield
4a	H	2a	72%
4b	6-H ₃ C	2b	72%
4c	6-F	2c	71%
4d	7-H ₃ C	2d	73%
4e	7-H ₃ CO	2e	66%
4f	7-F	2f	52%
4g	8-H ₃ C	2g	75%

Chemical reaction scheme showing the conversion of compound **2f** to compound **7**. Compound **2f** (a quinoline derivative with a CF_3 group at position 8, a fluorine at position 6, and a bromine at position 4) reacts to form a lithium salt intermediate (shown in brackets), which then reacts to form compound **7** (a quinoline derivative with a CF_3 group at position 8, a fluorine at position 6, and a carboxylic acid group (HOOC) at position 4).

Table 5. 2-Bromo-4-trifluoromethyl-3-quinolinecarboxylic acids **5** by consecutive treatment of 2-bromoquinolines **2** with lithium diisopropylamide and carbon dioxide

Compound	R	Precursor	Yield
5a	H	2a	81%
5b	6-H ₃ C	2b	66%
5c	6-F	2c	69%
5d	7-H ₃ C	2d	88%
5e	7-H ₃ CO	2e	82%
5f	7-F	2f	0% ^[a]
5g	8-H ₃ C	2g	89%

^[a] Acid **7** (87%) instead (see text).

A few of the acids **5** were subjected to reductive debromination (Table 6). This was accomplished either by bromine/lithium permutation and subsequent quenching with methanol or by catalytic hydrogenation. In addition, the 2-formyl-8-methyl-4-trifluoromethyl-3-carboxylic acid (**8**; 57%) was obtained from the halogen/metal permutation performed with the carboxylate of the bromo acid **5g** followed by addition to *N,N*-dimethylformamide and hydrolysis.

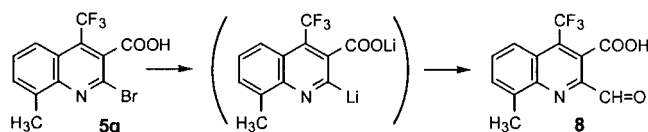


Table 6. 4-Trifluoromethyl-3-quinolinecarboxylic acids **6** by two-step debromination of the bromo analogs **5**

Compound	R	Precursor	Yield
6a	H	5a	64% ^[a]
6c	6-F	5c	71% ^[b]
6g	8-H ₃ C	5g	67% ^[a]

^[a] After consecutive treatment with butyllithium and methanol. ^[b] Bromine removal by catalytic hydrogenation.

This report has to close with a word of warning. There is no a priori certainty about the structural identity of the aniline/ β -oxo ester condensation/cyclization products. A minor change in the protocol (e.g. the use of methanesulfonic acid in the presence of phosphorus pentoxide^[8] rather than concentrated sulfuric acid) suffices to switch back from the 4-trifluoromethyl-2-quinolone to the 2-trifluoromethyl-4-quinoline family. To avoid erroneous assignments, a careful evaluation of the spectroscopic data is warranted. Characteristic differences between the two isomeric series are displayed by the ¹H NMR spectra of the 4- or 2-(trifluoromethyl)quinolines (**3** and **10**, respectively: $\delta \approx 8.3$ vs. 9.0 ppm; see Table 7) and, even more markedly, by the ¹³C NMR spectra of the 4- or 2-trifluoromethyl-2- or -4-quinolinones (**1** and **9**, respectively: $\delta \approx 149$ vs. 137 ppm; see Table 8).

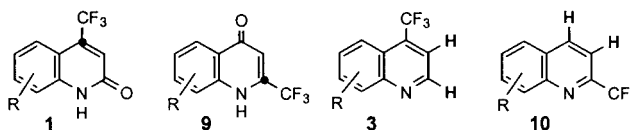


Table 7. The ¹H NMR shifts (in ppm relative to tetramethylsilane) at the 2- and 3-position of representative 4-trifluoromethyl-2-quinolines **3** and at the 3- and 4-position of the isomeric 2-(trifluoromethyl)quinolines **10**

R	Compound	$\delta(2\text{-H})^{\text{[a]}}$	$\delta(3\text{-H})^{\text{[b]}}$	Compound	$\delta(3\text{-H})^{\text{[a]}}$	$\delta(4\text{-H})^{\text{[a]}}$
H	3a	9.05	7.70	10a	8.37	7.75
6-H ₃ C	3b	8.97	7.70	10b	8.26	7.70
6-F	3c	9.00	7.72	10c	8.30	7.75
7-H ₃ C	3d	8.99	7.62	10d	—	—
7-H ₃ CO	3e	8.94	7.54	10e	8.21	4.58
7-F	3f	9.06	7.69	10f	8.35	7.73
8-H ₃ C	3g	9.01	7.66	10g	—	—

^[a] ³*J* = 8.3–8.7 Hz. ^[b] ³*J* = 4.0–4.7 Hz.

Table 8. ¹³C NMR shifts (in ppm relative to tetramethylsilane) and coupling constants (in Hz) of C-4 and C-2 of representative 4-trifluoromethyl-2-quinolones **1** and 2-trifluoromethyl-4-quinolones **9**, respectively

R	Compound	$\delta(\text{CF}_3)$	² <i>J</i> _{C,F}	Compound	$\delta(\text{CF}_3)$	² <i>J</i> _{C,F}
H	1a	136.6	34.5	9a	147.7	34.4
6-H ₃ C	1b	137.3	34.0	9b	147.8	35.3
6-F	1c	135.7	32.7	9c	146.2	32.9
7-H ₃ C	1d	136.2	31.2	9d	—	—
7-H ₃ CO	1e	137.5	30.8	9e	149.0	35.4
7-F	1f	137.4	31.4	9f	150.2	34.4
8-H ₃ C	1g	137.9	30.6	9g	—	—

Experimental Section

Generalities: Working practices and abbreviations are specified in previous articles from this laboratory.^[10–12] ¹H and ¹³C NMR spectra were recorded at 400 and 101 MHz relative to the internal standard tetramethylsilane ($\delta = 0.00$ ppm), samples having been dissolved in deuteriochloroform or, if marked by an asterisk (*), in perdeuterioacetone.

4-Trifluoromethyl-Substituted 2-Quinolinones **1**

4-Trifluoromethyl-2(1*H*)-quinolinone (1a): Aniline (9.1 mL, 9.3 g, 0.10 mol) and ethyl 4,4,4-trifluoro-3-oxobutanoate (29 mL, 37 g, 0.20 mol) were heated at 110 °C for 45 min before the excess of ester was evaporated under reduced pressure. Then 75% aqueous sulfuric acid (80 mL) was added to the oily mass formed. The mixture was heated for 45 min at 90 °C, then poured on milled ice (0.25 L). The precipitate formed was collected by filtration and crystallized from ethanol; colorless needles; m.p. 246–247 °C (ref.^[9] 245–246 °C; reprod.); yield: 7.25 g (34%). ¹H NMR (D₃CSOCD₃): $\delta = 7.71$ (d, *J* = 8.4 Hz, 1 H), 7.65 (td, *J* = 7.6,

1.0 Hz, 1 H), 7.43 (d, $J = 8.5$ Hz, 1 H), 7.32 (td, $J = 7.7$, 1.0 Hz, 1 H), 6.98 (s, 1 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 160.0$, 139.8, 136.6 (q, $J = 31$ Hz), 131.7, 124.3, 122.8, 122.6 (q, $J = 277$ Hz), 122.0 (q, $J = 5$ Hz), 116.5, 113.1 ppm. MS (c.i.): m/z (%) = 231 (27) [$\text{M}^+ + 18$], 214 (100) [$\text{M}^+ + 1$], 213 (34) [M^+], 185 (8), 166 (2). $\text{C}_{10}\text{H}_6\text{F}_3\text{NO}$ (213.16): calcd. C 56.35, H 2.84; found C 56.19, H 3.00.

6-Methyl-4-trifluoromethyl-2(1H)-quinolinone (1b): Analogously, using *p*-toluidine (11 g, 0.10 mol); colorless needles; m.p. 257–269 °C (from ethanol; reprod.); yield: 7.27 g (32%). ^1H NMR (D_3CSOCD_3): $\delta = 7.6$ (m, 2 H), 7.47 (d, $J = 8.5$ Hz, 1 H), 7.07 (s, 1 H), 2.54 (s, 3 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 160.9$, 138.9, 137.3 (q, $J = 33$ Hz), 134.9, 133.3, 133.1, 123.7, 123.5 (q, $J = 277$ Hz), 117.4, 114.0, 21.7 ppm. MS (c.i.): m/z (%) = 246 (0) [$\text{M}^+ + 18$], 228 (100) [$\text{M}^+ + 1$], 198 (4). $\text{C}_{11}\text{H}_8\text{F}_3\text{NO}$ (227.15): calcd. C 58.16, H 3.55, N 6.17; found C 58.09, H 3.55, N 6.12.

6-Fluoro-4-trifluoromethyl-2(1H)-quinolinone (1c): Analogously, using *p*-fluoroaniline (9.5 mL, 11 g, 0.10 mol); colorless needles; m.p. 252–253 °C (from ethanol; decomp.); yield: 2.54 g (11%). ^1H NMR (D_3CSOCD_3): $\delta = 7.59$ (td, $J = 9.2$, 2.6 Hz, 1 H), 7.49 (dd, $J = 9.1$, 4.9 Hz, 1 H), 7.41 (d, $J = 9.5$ Hz, 1 H), 7.08 (s, 1 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 159.6$, 156.4 (d, $J = 238$ Hz), 136.5, 135.7 (q, $J = 34$ Hz), 123.1 (d, $J = 5$ Hz), 121.9 (q, $J = 275$ Hz), 120.0 (d, $J = 25$ Hz), 118.4 (d, $J = 18$ Hz), 113.4 (d, $J = 9$ Hz), 109.3 (d, $J = 25$ Hz) ppm. MS (c.i.): m/z (%) = 249 (0) [$\text{M}^+ + 18$], 232 (40) [$\text{M}^+ + 1$], 231 (100) [M^+], 203 (26), 184 (19), 107 (9). $\text{C}_{10}\text{H}_5\text{F}_4\text{NO}$ (231.15): calcd. C 51.96, H 2.18, N 6.06; found C 52.03, H 2.10, N 6.11.

7-Methyl-4-trifluoromethyl-2(1H)-quinolinone (1d): Analogously, using *m*-toluidine (11 mL, 11 g, 0.10 mol); colorless needles; m.p. 210–213 °C (from ethanol; reprod.); yield: 6.59 g (29%). ^1H NMR (D_3CSOCD_3): $\delta = 7.59$ (d, $J = 8.6$ Hz, 1 H), 7.22 (s, 1 H), 7.14 (d, $J = 8.2$ Hz, 1 H), 6.89 (s, 1 H), 2.41 (s, 3 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 160.1$, 142.0, 139.7, 136.2 (q, $J = 31$ Hz), 124.2, 123.8, 122.5 (q, $J = 276$ Hz), 120.5 (q, $J = 5$ Hz), 116.0, 110.9, 21.1 ppm. MS (c.i.): m/z (%) = 245 (0) [$\text{M}^+ + 18$], 228 (100) [$\text{M}^+ + 1$], 198 (7), 130 (3), 102 (1). $\text{C}_{11}\text{H}_8\text{F}_3\text{NO}$ (227.15): calcd. C 58.16, H 3.55, N 6.17; found C 58.06, H 3.64, N 6.31.

7-Methoxy-4-trifluoromethyl-2(1H)-quinolinone (1e): Analogously, using *p*-anisidine (11 mL, 12 g, 0.10 mol); tiny yellowish needles; m.p. 219–222 °C (from ethanol; reprod.); yield: 7.29 g (30%). ^1H NMR (D_3CSOCD_3): $\delta = 7.72$ (dm, $J = 8.0$ Hz, 1 H), 7.05 (dd, $J = 9.5$, 2.5 Hz, 1 H), 7.04 (s, 1 H), 6.89 (s, 1 H), 3.94 (s, 3 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 162.9$, 161.7, 143.0, 137.5 (q, $J = 31$ Hz), 126.8, 123.8 (q, $J = 278$ Hz), 119.4 (q, $J = 6$ Hz), 113.3, 108.1, 100.2, 56.7 ppm. MS (c.i.): m/z (%) = 261 (0) [$\text{M}^+ + 18$], 244 (100) [$\text{M}^+ + 1$]. $\text{C}_{11}\text{H}_8\text{F}_3\text{NO}_2$ (243.14): calcd. C 54.34, H 3.32, N 5.76; found C 54.46, H 3.32, N 5.88.

7-Fluoro-4-trifluoromethyl-2(1H)-quinolinone (1f): Analogously, using *m*-fluoroaniline (11 mL, 11 g, 0.10 mol); tiny colorless needles; m.p. 166–167 °C (from ethanol; decomp.); yield: 7.16 g (31%). ^1H NMR (D_3CSOCD_3): $\delta = 7.87$ (tm, $J = 8.0$ Hz, 1 H), 7.30 (m, 2 H), 7.05 (s, 1 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 164.2$ (d, $J = 247$ Hz), 161.2, 142.6 (d, $J = 12$ Hz), 137.4 (d, $J = 31$ Hz), 128.0 (d, $J = 10$ Hz), 123.2 (q, $J = 267$ Hz), 121.5, 112.4 (d, $J = 23$ Hz), 111.2, 103.1 (d, $J = 26$ Hz) ppm. MS (c.i.): m/z (%) = 249 (0) [$\text{M}^+ + 18$], 232 (100) [$\text{M}^+ + 1$], 231 (88) [M^+], 203 (19), 184 (8). $\text{C}_{10}\text{H}_5\text{F}_4\text{NO}$ (231.10): calcd. C 51.97, H 2.18, N 6.06; found C 52.03, H 1.95, N 6.18.

8-Methyl-4-trifluoromethyl-2(1H)-quinolinone (1g): Analogously using *o*-toluidine (11 mL, 11 g, 0.10 mol); tiny colorless needles;

m.p. 248–251 °C (from ethanol; reprod.); yield: 4.77 g (21%). ^1H NMR (D_3CSOCD_3): $\delta = 7.68$ (d, $J = 7.3$ Hz, 1 H), 7.60 (d, $J = 7.0$ Hz, 1 H), 7.34 (t, $J = 7.6$ Hz, 1 H), 7.11 (s, 1 H), 2.61 (s, 3 H) ppm. ^{13}C NMR (D_3CSOCD_3): $\delta = 161.3$, 139.0, 137.9 (q, $J = 31$ Hz), 133.8, 125.9, 123.7 (q, $J = 272$ Hz), 123.5, 123.0, 122.6, 114.1, 18.8 ppm. MS (c.i.): m/z (%) = 245 (0) [$\text{M}^+ + 18$], 228 (100) [$\text{M}^+ + 1$], 227 (35), 198 (2), 130 (1). $\text{C}_{11}\text{H}_8\text{F}_3\text{NO}$ (227.15): calcd. C 58.16, H 3.55, N 6.17; found C 57.93, H 3.29, N 6.18.

4-Trifluoromethyl-Substituted 2-Bromoquinolines 2

2-Bromo-4-(trifluoromethyl)quinoline (2a): At 60 °C, quinolinone **1a** (11 g, 50 mmol) was added to an excess of phosphorus oxybromide (29 g, 0.10 mol). The mixture was heated at 140 °C for 2 h before being poured onto ice (0.20 L). The precipitated brown mass was collected and dissolved in methanol (0.20 L) containing charcoal (3 g). The slurry was stirred at 25 °C for 1 h before being filtered through a pad of diatomaceous earth. The residue left after evaporation of the solvent was crystallized from methanol; colorless needles; m.p. 56–57 °C (after sublimation); yield: 9.11 g (66%). ^1H NMR: $\delta = 8.15$ (d, $J = 8.5$ Hz, 1 H), 8.11 (d, $J = 8.5$ Hz, 1 H), 7.83 (m, 2 H), 7.71 (ddd, $J = 8.4$, 7.0, 1.3 Hz, 1 H) ppm. ^{13}C NMR: $\delta = 149.5$, 140.6, 136.2 (q, $J = 33$ Hz), 131.4, 129.6, 128.6, 124.1, 123.1 (q, $J = 6$ Hz), 122.5 (q, $J = 277$ Hz), 121.9 ppm. MS (c.i.): m/z (%) = 294 (0) [$\text{M}^+ + 18$], 277 (100) [$\text{M}^+ + 1$], 276 (82) [M^+], 213 (14), 196 (67), 176 (47). $\text{C}_{10}\text{H}_5\text{BrF}_3\text{N}$ (276.05): calcd. C 43.51, H 1.83; found C 43.43, H 1.71.

2-Bromo-6-methyl-4-(trifluoromethyl)quinoline (2b): Analogously, from quinolinone **1b** (11 g, 50 mmol); colorless needles; m.p. 97–99 °C (from methanol); yield: 8.56 g (59%). ^1H NMR: $\delta = 8.03$ (d, $J = 8.5$ Hz, 1 H), 7.83 (s, 1 H), 7.76 (s, 1 H), 7.65 (dd, $J = 8.5$, 1.7 Hz, 1 H), 2.57 (s, 3 H) ppm. ^{13}C NMR: $\delta = 147.8$, 139.3, 139.2, 135.7 (qr, $J = 32$ Hz), 133.6, 129.0, 122.9, 122.4 (qr, $J = 276$ Hz), 121.7, 22.0 ppm. MS (c.i.): m/z (%) = 308 (0) [$\text{M}^+ + 18$], 291 (38) [$\text{M}^+ + 1$], 290 (100) [M^+], 227 (11), 212 (5). $\text{C}_{11}\text{H}_7\text{BrF}_3\text{N}$ (290.06): calcd. C 44.55, H 2.43, N 4.83; found C 45.15, H 2.32, N 4.89.

2-Bromo-6-fluoro-4-(trifluoromethyl)quinoline (2c): Analogously, from quinolinone **1c** (12 g, 50 mmol); colorless needles; m.p. 108–109 °C (from methanol); yield: 8.67 g (59%). ^1H NMR: $\delta = 8.16$ (dd, $J = 9.3$, 5.4 Hz, 1 H), 7.85 (s, 1 H), 7.73 (dm, $J = 9.4$ Hz, 1 H), 7.61 (ddd, $J = 9.2$, 7.9, 2.7 Hz, 1 H) ppm. ^{13}C NMR: $\delta = 161.4$ (d, $J = 257$ Hz), 146.5, 139.8, 136.1 (qr, $J = 35$ Hz), 132.2 (qr, $J = 5$ Hz), 124.0 (d, $J = 25$ Hz), 122.8, 122.2 (qr, $J = 275$ Hz), 121.8 (d, $J = 26$ Hz), 108.7 (d, $J = 25$ Hz) ppm. MS (c.i.): m/z (%) = 312 (0) [$\text{M}^+ + 18$], 295 (62) [$\text{M}^+ + 1$], 294 (100) [M^+], 276 (2), 214 (5), 145 (2). $\text{C}_{10}\text{H}_4\text{BrF}_4\text{N}$ (294.04): calcd. C 40.85, H 1.37; found C 40.82, H 1.81.

2-Bromo-7-methyl-4-(trifluoromethyl)quinoline (2d): Analogously, from quinolinone **1d** (11 g, 50 mmol); colorless needles; m.p. 49–51 °C (from methanol); yield: 8.99 g (62%). ^1H NMR: $\delta = 8.00$ (d, $J = 8.9$ Hz, 1 H), 7.92 (s, 1 H), 7.75 (s, 1 H), 7.55 (d, $J = 8.9$ Hz, 1 H), 2.60 (s, 3 H) ppm. ^{13}C NMR: $\delta = 149.7$, 142.3, 140.7, 136.2 (qr, $J = 32$ Hz), 131.0, 128.7, 123.7, 122.7 (qr, $J = 276$ Hz), 122.1, 120.9, 21.6 ppm. MS (c.i.): m/z (%) = 308 (0) [$\text{M}^+ + 18$], 290 (100) [M^+], 227 (7), 190 (3). $\text{C}_{11}\text{H}_7\text{BrF}_3\text{N}$ (290.06): calcd. C 45.55, H 2.43, N 4.83; found C 45.55, H 2.23, N 4.90.

2-Bromo-7-methoxy-4-(trifluoromethyl)quinoline (2e): Analogously, from quinolinone **1e** (12 g, 50 mmol); colorless needles; m.p. 76–78 °C (from methanol); yield: 8.11 g (53%). ^1H NMR: $\delta = 7.96$ (dd, $J = 9.2$, 2.0 Hz, 1 H), 7.66 (s, 1 H), 7.44 (d, $J = 2.3$ Hz, 1 H), 7.31 (dd, $J = 9.2$, 2.8 Hz, 1 H), 3.95 (s, 3 H) ppm. ^{13}C NMR: $\delta = 161.9$, 151.4, 141.0, 136.1 (qr, $J = 32$ Hz), 125.1, 122.5 (qr, $J = 282$ Hz),

121.7, 120.6, 116.8, 107.7, 55.8 ppm. MS (c.i.): m/z (%) = 324 (0) [$M^+ + 18$], 307 (40) [$M^+ + 1$], 306 (100) [M^+]. $C_{11}H_7BrF_3NO$ (306.05): calcd. C 43.17, H 2.31, N 4.58; found C 43.06, H 2.51, N 4.56.

2-Bromo-7-fluoro-4-(trifluoromethyl)quinoline (2f): Analogously, from quinolinone **1f** (12 g, 50 mmol); colorless needles; m.p. 60–62 °C (from methanol); yield: 7.50 g (51%). 1H NMR: δ = 8.13 (dq, J = 7.8, 2.0 Hz, 1 H), 7.81 (s, 1 H), 7.78 (dd, J = 9.3, 2.9 Hz, 1 H), 7.51 (ddd, J = 9.5, 8.1, 2.9 Hz, 1 H) ppm. ^{13}C NMR: δ = 163.9 (d, J = 256 Hz), 150.5 (d, J = 13 Hz), 142.1, 136.2 (qr, J = 33 Hz), 127.1 (d, J = 10 Hz), 122.5, 122.1 (qr, J = 276 Hz), 119.1 (d, J = 26 Hz), 119.0, 113.8 (d, J = 21 Hz) ppm. MS (c.i.): m/z (%) = 312 (0) [$M^+ + 18$], 295 (100) [$M^+ + 1$], 226 (14), 194 (7), 145 (18). $C_{10}H_4BrF_4N$ (294.01): calcd. C 40.85, H 1.37, N 4.76; found C 40.99, H 1.25, N 4.94.

2-Bromo-8-methyl-4-(trifluoromethyl)quinoline (2g): Analogously, from quinolinone **1g** (11 g, 50 mmol); colorless needles; m.p. 17–19 °C (from methanol); b.p. 90–93 °C/5 Torr; yield: 9.28 g (64%). 1H NMR: δ = 7.94 (d, J = 8.2 Hz, 1 H), 7.81 (s, 1 H), 7.67 (d, J = 7.1 Hz, 1 H), 7.59 (dd, J = 8.2, 7.3 Hz, 1 H), 2.78 (s, 3 H) ppm. ^{13}C NMR: δ = 148.4, 139.4, 137.8, 136.4 (qr, J = 32 Hz), 134.4, 128.3, 122.8 (qr, J = 6 Hz), 122.6 (qr, J = 276 Hz), 122.0, 121.9, 15.8 ppm. MS (c.i.): m/z (%) = 318 (0) [$M^+ + 18$], 292 (100) [$M^+ + 2$], 291 (39) [$M^+ + 1$], 227 (2), 99 (2). $C_{11}H_7BrF_3N$ (290.06): calcd. C 45.55, H 2.43, N 4.83; found C 45.59, H 2.18, N 4.90.

4-Trifluoromethyl-Substituted Quinolines 3

4-(Trifluoromethyl)quinoline (3a): 2-Bromo-4-(trifluoromethyl)-quinoline **2a** (2.8 g, 10 mmol) was added to butyllithium (10 mmol) in toluene (15 mL) and hexanes (6.0 mL), cooled in a methanol/dry ice bath. After 15 min at –75 °C, methanol (1.0 mL, 0.79 g, 25 mmol) was injected. The product was isolated by direct distillation; colorless liquid; b.p. 135–137 °C/23 Torr (ref.^[9] 139 °C); n_D^{20} = 1.5375; d_4^{20} = 1.364; yield: 1.24 g (63%). 1H NMR: δ = 9.05 (d, J = 4.2 Hz, 1 H), 8.23 (d, J = 8.5 Hz, 1 H), 8.12 (dm, J = 8.5 Hz, 1 H), 7.83 (ddd, J = 8.5, 6.9, 1.4 Hz, 1 H), 7.70 (m, 2 H) ppm. ^{13}C NMR: δ = 149.5, 149.0, 134.4 (qr, J = 32 Hz), 130.4 (2 C), 128.5, 124.0, 123.4 (qr, J = 275 Hz), 123.0, 118.0 ppm. MS (c.i.): m/z (%) = 215 (0) [$M^+ + 18$], 198 (12) [$M^+ + 1$], 197 (51) [M^+], 175 (5), 157 (3), 83 (100).

6-Methyl-4-(trifluoromethyl)quinoline (3b): Analogously, from the 2-bromoquinoline **2b** (2.9 g, 10 mmol); colorless needles; m.p. 37–39 °C (from hexanes); yield: 1.53 g (72%). 1H NMR: δ = 8.97 (d, J = 4.4 Hz, 1 H), 8.13 (d, J = 8.7 Hz, 1 H), 7.90 (s, 1 H), 7.7 (m, 2 H) ppm. ^{13}C NMR: δ = 148.3, 147.3, 138.7, 133.7 (qr, J = 32 Hz), 132.7, 130.0, 123.4 (qr, J = 275 Hz), 123.1, 122.8, 117.9 (qr, J = 6 Hz), 22.0 ppm. MS (c.i.): m/z (%) = 229 (0) [$M^+ + 18$], 212 (24) [$M^+ + 1$], 211 (100) [M^+], 192 (4), 142 (7). $C_{11}H_8F_3N$ (211.19): calcd. C 62.56, H 3.82, N 6.63; found C 62.46, H 3.82, N 6.67.

6-Fluoro-4-(trifluoromethyl)quinoline (3c): Analogously, from 2-bromoquinoline **2c** (2.9 g, 10 mmol); colorless needles; m.p. 51–53 °C (from hexanes); yield: 1.45 g (68%). 1H NMR: δ = 9.00 (d, J = 4.4 Hz, 1 H), 8.22 (dd, J = 9.1, 5.8 Hz, 1 H), 7.76 (dm, J = 9.8 Hz, 1 H), 7.72 (d, J = 4.2 Hz, 1 H), 7.59 (ddd, J = 12.1, 10.3, 3.8 Hz, 1 H) ppm. ^{13}C NMR: δ = 161.3 (d, J = 252 Hz), 148.7 (d, J = 2 Hz), 146.0, 133.9 (qd, J = 32, 6 Hz), 132.9 (d, J = 10 Hz), 123.7 (d, J = 10 Hz), 123.2 (qr, J = 275 Hz), 120.6 (d, J = 26 Hz), 118.6 (d, J = 5 Hz), 108.0 (d, J = 23 Hz) ppm. MS (c.i.): m/z (%) = 233 (1) [$M^+ + 18$], 216 (85) [$M^+ + 1$], 215 (5) [M^+], 170 (12), 122 (40),

99 (100). $C_{10}H_5F_4N$ (215.15): calcd. C 55.83, H 2.34; found C 55.57, H 2.20.

7-Methyl-4-(trifluoromethyl)quinoline (3d): Analogously, from the 2-bromoquinoline **2d** (2.9 g, 10 mmol); colorless needles; m.p. 48–50 °C (from hexanes); yield: 1.50 g (71%). 1H NMR: δ = 8.99 (d, J = 4.5 Hz, 1 H), 8.04 (dq, J = 8.8, 2.0 Hz, 1 H), 8.00 (qr, J = 0.8 Hz, 1 H), 7.62 (d, J = 4.2 Hz, 1 H), 7.54 (dd, J = 9.0, 2.0 Hz, 1 H), 2.61 (s, 3 H) ppm. ^{13}C NMR: δ = 149.5, 149.2, 140.6, 133.8 (qr, J = 32 Hz), 130.6, 129.2, 124.0, 123.9 (qr, J = 275 Hz), 120.8, 117.2, 21.8 ppm. MS (c.i.): m/z (%) = 229 (0) [$M^+ + 18$], 212 (22) [$M^+ + 1$], 211 (100) [M^+], 192 (4), 142 (9), 115 (4). $C_{11}H_8F_3N$ (211.19): calcd. C 62.56, H 3.82; found C 62.29, H 3.94.

7-Methoxy-4-(trifluoromethyl)quinoline (3e): Analogously, from the 2-bromoquinoline **2e** (3.1 g, 10 mmol); colorless needles; m.p. 41–43 °C (from hexanes); yield: 1.55 g (68%). 1H NMR: δ = 8.03 (d, J = 4.3 Hz, 1 H), 7.54 (d, J = 8.9 Hz, 1 H), 7.52 (s, 1 H), 7.33 (d, J = 4.5 Hz, 1 H), 3.99 (s, 3 H) ppm. ^{13}C NMR: δ = 161.0, 150.7, 148.7, 134.2 (qr, J = 32 Hz), 125.1, 123.3 (qr, J = 276 Hz), 121.6, 118.0, 115.6, 108.1, 55.7 ppm. MS (c.i.): m/z (%) = 262 (0) [$M^+ + 18$], 245 (0) [$M^+ + 1$], 227 (100) [M^+], 208 (4), 197 (19), 147 (5). $C_{11}H_8F_3NO$ (227.18): calcd. C 58.16, H 3.55, N 6.17; found C 58.00, H 3.48, N 6.08.

7-Fluoro-4-(trifluoromethyl)quinoline (3f): Analogously, from the 2-bromoquinoline **2f** (2.9 g, 10 mmol); colorless needles; m.p. 62–64 °C (from hexanes); yield: 1.17 g (55%). 1H NMR: δ = 9.06 (d, J = 4.3 Hz, 1 H), 8.17 (tq, J = 8.0, 1.9 Hz, 1 H), 7.86 (dd, J = 7.0, 2.6 Hz, 1 H), 7.69 (d, J = 4.7 Hz, 1 H), 7.50 (ddd, J = 9.6, 8.0, 2.8 Hz, 1 H) ppm. ^{13}C NMR: δ = 163.1 (d, J = 253 Hz), 150.7, 150.3 (qr, J = 13 Hz), 134.6 (qr, J = 32 Hz), 126.4 (d, J = 9 Hz), 123.3 (qr, J = 275 Hz), 120.0, 118.9 (d, J = 26 Hz), 117.3, 114.1 (d, J = 20 Hz) ppm. $C_{10}H_5F_4N$ (215.15): calcd. C 53.83, H 2.34, N 6.51; found C 53.71, H 2.49, N 6.51.

8-Methyl-4-(trifluoromethyl)quinoline (3g): Analogously, from the 2-bromoquinoline **2g** (2.9 g, 10 mmol); colorless needles; m.p. 24–25 °C (from hexanes); yield: 1.56 g (74%). 1H NMR: δ = 9.05 (d, J = 4.4 Hz, 1 H), 7.99 (d, J = 8.5 Hz, 1 H), 7.70 (d, J = 4.4 Hz, 1 H), 7.66 (d, J = 6.9 Hz, 1 H), 7.57 (dd, J = 8.4, 7.5 Hz, 1 H), 2.84 (s, 3 H) ppm. ^{13}C NMR: δ = 148.1, 148.0, 138.0, 134.1 (qr, J = 32 Hz), 130.2, 127.8, 123.3 (qr, J = 275 Hz), 122.7, 121.6, 98.9 (qr, J = 5 Hz), 17.9 ppm. $C_{11}H_8F_3N$ (211.19): calcd. C 62.56, H 3.82, N 6.63; found C 62.54, H 3.67, N 6.51.

4-Trifluoromethyl-Substituted 2-Quinolincarboxylic Acids 4

4-Trifluoromethyl-2-quinolincarboxylic Acid (4a): The 2-bromoquinoline **2a** (4.1 g, 15 mmol) was dissolved in a solution of butyllithium (15 mmol) in tetrahydrofuran (30 mL) and hexanes (10 mL) cooled in a methanol/dry ice bath. After 15 min at –75 °C, the mixture was poured onto an excess of freshly crushed solid carbon dioxide. At +25 °C, tetrafluoroboric acid–diethyl ether (2.7 mL, 3.2 g, 20 mmol) was added with shaking and the volatiles were evaporated. The residue was crystallized from ethanol to afford colorless prisms; m.p. 142–143 °C; yield: 2.50 g (69%). 1H NMR*: δ = 8.47 (s, 1 H), 8.37 (ddd, J = 8.4, 1.43, 0.7 Hz, 1 H), 8.27 (d, J = 8.5 Hz, 1 H), 8.06 (td, J = 6.9, 1.4 Hz, 1 H), 7.99 (td, J = 7.0, 1.5 Hz, 1 H) ppm. ^{13}C NMR*: δ = 165.0, 158.5 (2 C), 139.0, 136.0 (qr, J = 35 Hz), 132.5, 131.8 (2 C), 124.7, 124.5 (qr, J = 274 Hz), 118.3 ppm. MS (c.i.): m/z (%) = 259 (0) [$M^+ + 18$], 242 (23) [$M^+ + 1$], 241 (22) [M^+], 197 (100), 176 (16), 147 (4), 128 (12). $C_{11}H_6F_3NO_2$ (241.17): calcd. C 54.78, H 2.51; found C 54.52, H 2.48.

6-Methyl-4-trifluoromethyl-2-quinolinecarboxylic Acid (4b): Analogously, from the 2-bromoquinoline **2b** (4.4 g, 15 mmol); colorless prisms; m.p. 205–208 °C (from ethanol; reprod.); yield: 2.77 g (72%). ¹H NMR*: δ = 8.44 (s, 1 H), 8.24 (d, *J* = 8.7 Hz, 1 H), 8.03 (s, 1 H), 7.90 (dd, *J* = 8.7, 1.7 Hz, 1 H), 2.68 (s, 3 H) ppm. ¹³C NMR*: δ = 165.2, 147.4, 142.8, 135.8, 135.2 (qr, *J* = 32 Hz), 134.7, 131.6, 124.8, 124.3 (qr, *J* = 275 Hz), 123.4, 118.4, 22.7 ppm. MS (c.i.): *m/z* (%) = 273 (0) [*M*⁺ + 18], 256 (43) [*M*⁺ + 1], 211 (100), 190 (65), 140 (64). C₁₂H₈F₃NO₂ (255.19): calcd. C 56.48, H 3.16, N 5.49; found C 56.51, H 3.33, N 5.52.

6-Fluoro-4-trifluoromethyl-2-quinolinecarboxylic Acid (4c): Analogously, from the 2-bromoquinoline **2c** (4.4 g, 15 mmol); colorless prisms; m.p. 168–171 °C (from ethanol; decomp.); yield: 2.76 g (71%). ¹H NMR*: δ = 8.52 (s, 1 H), 8.44 (dd, *J* = 9.3, 5.6 Hz, 1 H), 7.93 (ddd, *J* = 9.3, 8.2, 2.7 Hz, 1 H), 7.88 (dm, *J* = 9.8 Hz, 1 H) ppm. ¹³C NMR*: δ = 165.9, 164.4 (d, *J* = 253 Hz), 149.0, 146.8, 136.1 (m, 2 C), 126.5 (d, *J* = 11 Hz), 124.9 (qr, *J* = 274 Hz), 123.5 (d, *J* = 27 Hz), 120.0, 109.5 (d, *J* = 25 Hz) ppm. MS (c.i.): *m/z* (%) = 277 (0) [*M*⁺ + 18], 260 (41) [*M*⁺ + 1], 259 (22) [*M*⁺], 242 (13), 215 (100), 194 (23). C₁₁H₅F₄NO₂ (259.16): calcd. C 50.98, H 1.94, N 5.40; found C 50.91, H 1.91, N 5.44.

7-Methyl-4-trifluoromethyl-2-quinolinecarboxylic Acid (4d): Analogously, from the 2-bromoquinoline **2d** (4.4 g, 15 mmol); colorless prisms; m.p. 145–148 °C (from ethanol; reprod.); yield: 2.80 g (73%). ¹H NMR*: δ = 8.42 (s, 1 H), 8.16 (dq, *J* = 8.9, 2.1 Hz, 1 H), 8.12 (qr, *J* = 0.8 Hz, 1 H), 7.85 (dd, *J* = 8.5, 1.7 Hz, 1 H), 2.66 (s, 3 H) ppm. ¹³C NMR*: δ = 165.2, 149.0, 148.4, 143.4, 135.8 (qr, *J* = 32 Hz), 137.3, 131.2, 124.5 (qr, *J* = 275 Hz), 124.1, 122.7, 117.4 (qr, *J* = 5 Hz), 21.7 ppm. MS (c.i.): *m/z* (%) = 273 (0) [*M*⁺ + 18], 256 (100) [*M*⁺ + 1], 211 (11). C₁₂H₈F₃NO₂ (255.19): calcd. C 56.48, H 3.16, N 5.49; found C 56.44, H 3.07, N 5.44.

7-Methoxy-4-trifluoromethyl-2-quinolinecarboxylic Acid (4e): Analogously, from the 2-bromoquinoline **2e** (4.6 g, 15 mmol); colorless prisms; m.p. 165–168 °C (from ethanol; reprod.); yield: 2.68 g (66%). ¹H NMR*: δ = 8.32 (s, 1 H), 8.17 (dq, *J* = 9.2, 2.3 Hz, 1 H), 7.66 (d, *J* = 2.9 Hz, 1 H), 7.61 (dd, *J* = 9.3, 2.9 Hz, 1 H), 4.07 (s, 3 H) ppm. MS (c.i.): *m/z* (%) = 271 (88) [*M*⁺], 227 (100). C₁₂H₈F₃NO₃ (271.19): calcd. C 53.15, H 2.97, N 5.17; found C 52.97, H 2.84, N 5.12.

7-Fluoro-4-trifluoromethyl-2-quinolinecarboxylic Acid (4f): Analogously, from the 2-bromoquinoline **2f** (4.4 g, 15 mmol); colorless prisms; m.p. 167–169 °C (from ethanol; decomp.); yield: 2.02 g (52%). ¹H NMR*: δ = 8.46 (s, 1 H), 8.37 (tq, *J* = 5.7, 1.9 Hz, 1 H), 8.01 (dd, *J* = 9.4, 2.7 Hz, 1 H), 7.87 (ddd, *J* = 9.4, 8.3, 2.7 Hz, 1 H) ppm. ¹³C NMR*: δ = 163.9, 163.5 (qr, *J* = 255 Hz), 149.2 (qr, *J* = 13 Hz), 135.3 (qr, *J* = 32 Hz), 126.7 (d, *J* = 10 Hz), 123.2 (qr, *J* = 275 Hz), 121.2 (d, *J* = 26 Hz), 120.9, 116.9, 113.8 (d, *J* = 21 Hz) ppm. MS (c.i.): *m/z* (%) = 277 (0) [*M*⁺ + 18], 259 (100) [*M*⁺], 242 (9), 214 (15), 194 (7). C₁₁H₅F₄NO₂ (259.16): calcd. C 50.98, H 1.94, N 5.40; found C 51.11, H 2.06, N 5.27.

8-Methyl-4-trifluoromethyl-2-quinolinecarboxylic Acid (4g): Analogously, from the 2-bromoquinoline **2g** (4.4 g, 15 mmol); colorless prisms; m.p. 168–171 °C (from ethanol; reprod.); yield: 2.87 g (75%). ¹H NMR*: δ = 8.51 (s, 1 H), 8.09 (d, *J* = 7.9 Hz, 1 H), 7.9 (m, 2 H), 2.90 (s, 3 H) ppm. ¹³C NMR*: δ = 165.1, 147.7, 147.1, 140.4, 136.2 (qr, *J* = 30 Hz), 132.4, 131.6, 124.9, 124.3 (qr, *J* = 275 Hz), 122.4, 118.0 (qr, *J* = 6 Hz), 18.4 ppm. MS (c.i.): *m/z* (%) = 273 (0) [*M*⁺ + 18], 256 (100) [*M*⁺ + 1]. C₁₂H₈F₃NO₂ (255.19): calcd. C 56.48, H 3.16, N 5.49; found C 56.80, H 2.93, N 5.52.

4-Trifluoromethyl-Substituted 2-Bromo-3-quinolinecarboxylic Acids 5

2-Bromo-4-trifluoromethyl-3-quinolinecarboxylic Acid (5a): Diisopropylamine (2.6 mL, 1.8 g, 18 mmol) and the 2-bromoquinoline **2a** (4.1 g, 15 mmol) were added consecutively to butyllithium (15 mmol) in tetrahydrofuran (30 mL) and hexanes (10 mL), kept in a methanol/dry ice bath. After 2 h at –75 °C, the mixture was carboxylated and worked up as described in the preceding section; colorless needles; m.p. 172–173 °C (from water; reprod.); yield: 3.88 g (81%). ¹H NMR*: δ = 8.22 (d, *J* = 8.6 Hz, 1 H), 8.15 (dd, *J* = 7.7, 0.8 Hz, 1 H), 8.02 (td, *J* = 7.0, 1.3 Hz, 1 H), 7.93 (td, *J* = 7.0, 1.4 Hz, 1 H) ppm. ¹³C NMR*: δ = 167.1, 150.3, 138.9, 134.0, 132.7 (qr, *J* = 33 Hz), 131.6, 131.0, 126.2 (2 C), 124.2 (qr, *J* = 278 Hz), 123.0 ppm. MS (c.i.): *m/z* (%) = 338 (0) [*M*⁺ + 18], 321 (96) [*M*⁺ + 1], 320 (100) [*M*⁺], 319 (88) [*M*⁺], 277 (20), 276 (22), 275 (21), 274 (21), 240 (29), 176 (12). C₁₁H₅BrF₃NO₂ (320.06): calcd. C 41.28, H 1.57; found C 41.12, H 1.70.

2-Bromo-6-methyl-4-trifluoromethyl-3-quinolinecarboxylic Acid (5b): Analogously, from the 2-bromoquinoline **2b** (4.4 g, 15 mmol); colorless needles; m.p. 218–220 °C (from water; reprod.); yield: 3.31 g (66%). ¹H NMR*: δ = 8.03 (d, *J* = 8.9 Hz, 1 H), 7.97 (broad s, 1 H), 7.85 (dd, *J* = 8.9, 1.9 Hz, 1 H), 2.63 (s, 3 H) ppm. ¹³C NMR*: δ = 166.0, 147.9, 141.4, 136.6, 135.2, 131.4 (qr, *J* = 32 Hz), 130.0, 124.1, 123.5 (qr, *J* = 277 Hz), 122.2, 22.1 ppm. MS (c.i.): *m/z* (%) = 352 (0) [*M*⁺ + 18], 334 (100) [*M*⁺], 271 (4), 254 (9). C₁₂H₇BrF₃NO₂ (334.09): calcd. C 43.14, H 2.11, N 4.19; found C 43.31, H 2.03, N 4.22.

2-Bromo-6-fluoro-4-trifluoromethyl-3-quinolinecarboxylic Acid (5c): Analogously, from the 2-bromoquinoline **2c** (4.4 g, 15 mmol); colorless needles; m.p. 196–198 °C (from water; reprod.); yield: 3.50 g (69%). ¹H NMR*: δ = 8.25 (dd, *J* = 9.3, 5.6 Hz, 1 H), 7.90 (ddd, *J* = 9.9, 8.1, 2.6 Hz, 1 H), 7.86 (dm, *J* = 9.9 Hz, 1 H) ppm. ¹³C NMR*: δ = 166.6, 163.6 (d, *J* = 251 Hz), 147.3, 138.0, 133.8 (d, *J* = 10 Hz), 132.6 (qr, *J* = 32 Hz), 131.5, 124.3 (2C), 124.1 (qr, *J* = 276 Hz), 110.5 (d, *J* = 26 Hz) ppm. MS (c.i.): *m/z* (%) = 356 (0) [*M*⁺ + 18], 339 (100) [*M*⁺ + 1], 338 (61) [*M*⁺], 337 (99) [*M*⁺], 293 (15), 258 (46), 194 (52). C₁₁H₄BrF₄NO₂ (338.05): calcd. C 39.08, H 1.19, N 4.14; found C 38.93, H 1.27, N 4.17.

2-Bromo-7-methyl-4-trifluoromethyl-3-quinolinecarboxylic Acid (5d): Analogously, from the 2-bromoquinoline **2d** (4.4 g, 15 mmol); colorless needles; m.p. 185–187 °C (from water; reprod.); yield: 4.41 g (88%). ¹H NMR*: δ = 8.13 (dq, *J* = 8.9, 2.5 Hz, 1 H), 7.96 (s, 1 H), 7.78 (dd, *J* = 9.2, 1.9 Hz, 1 H), 2.64 (s, 3 H) ppm. ¹³C NMR*: δ = 166.2, 149.6, 144.4, 137.9, 133.1, 132.1 (qr, *J* = 32 Hz), 129.3, 129.2, 125.3 (qr, *J* = 4 Hz), 123.6 (qr, *J* = 278 Hz), 120.2, 21.7 ppm. MS (c.i.): *m/z* (%) = 352 (0) [*M*⁺ + 18], 336 (100) [*M*⁺ + 2], 254 (12), 228 (7), 140 (2). C₁₂H₇BrF₃NO₂ (334.09): calcd. C 43.14, H 2.11, N 4.19; found C 43.12, H 1.88, N 4.19.

2-Bromo-7-methoxy-4-trifluoromethyl-3-quinolinecarboxylic Acid (5e): Analogously, from the 2-bromoquinoline **2e** (4.6 g, 15 mmol); colorless needles; m.p. 179–180 °C (from water; reprod.); yield: 4.30 g (82%). ¹H NMR*: δ = 8.13 (d, *J* = 9.6 Hz, 1 H), 7.53 (d, *J* = 10.0 Hz, 1 H), 7.52 (s, 1 H), 4.08 (s, 3 H) ppm. ¹³C NMR*: δ = 167.2, 163.2, 151.0, 138.8, 131.0 (qr, *J* = 32 Hz), 128.1, 126.7, 124.1, 123.4 (qr, *J* = 277 Hz), 116.8, 108.9, 57.4 ppm. MS (c.i.): *m/z* (%) = 368 (0) [*M*⁺ + 18], 351 (100) [*M*⁺ + 1], 332 (54), 270 (18), 225 (17). C₁₂H₇BrF₃NO₃ (350.04): calcd. C 41.17, H 2.02, N 4.00; found C 41.01, H 2.18, N 4.00.

2-Bromo-8-methyl-4-trifluoromethyl-3-quinolinecarboxylic Acid (5g): Analogously, from the 2-bromoquinoline **2g** (4.4 g, 15 mmol); colorless prisms; m.p. 178–181 °C (from water; reprod.); yield: 4.46 g (89%). ¹H NMR*: δ = 8.09 (d, *J* = 9.8 Hz, 1 H), 7.91 (d, *J* = 7.6 Hz, 1 H), 7.83 (dd, *J* = 8.4, 7.2 Hz, 1 H), 2.78 (s, 3 H)

ppm. ^{13}C NMR*: δ = 166.2, 148.2, 138.6, 136.7, 133.1, 132.3 (qr, J = 30 Hz), 130.6, 129.9 (qr, J = 3 Hz), 123.4 (qr, J = 278 Hz), 123.1 (qr, J = 3 Hz), 122.3, 18.2 ppm. MS (c.i.): m/z (%) = 352 (0) [M^+ + 18], 336 (100) [M^+ + 2], 335 (42) [M^+ + 1], 290 (23), 256 (19). $\text{C}_{12}\text{H}_7\text{BrF}_3\text{NO}_2$ (334.09): calcd. C 43.14, H 2.11, N 4.19; found C 43.19, H 1.84, N 4.33.

Trifluoromethyl-Substituted 3-Quinolinecarboxylic Acids 6

4-Trifluoromethyl-3-quinolinecarboxylic Acid (6a): A solution of the 2-bromo-3-quinolinecarboxylic acid **5a** (1.5 g, 5.0 mmol) in diethyl ether (15 mL) was added dropwise, over 15 min, to butyllithium (10 mmol) in diethyl ether (15 mL) and hexanes (6.0 mL), cooled to -100°C . The mixture was kept at this temperature for 2 h before being treated with methanol (1.0 mL, 0.80 g, 25 mmol) and worked up as described above (see the isolation of the acids **4**); colorless prisms; m.p. $182\text{--}184^\circ\text{C}$ (from acetic acid; decomp.); yield: 0.77 g (64%). ^1H NMR*: δ = 9.14 (s, 1 H), 8.27 (d, J = 8.1 Hz, 1 H), 8.26 (d, J = 8.7 Hz, 1 H), 8.00 (ddd, J = 8.2, 7.0, 1.2 Hz, 1 H), 7.90 (ddd, J = 8.9, 7.0, 1.6 Hz, 1 H) ppm. ^{13}C NMR*: δ = 166.9, 148.9, 147.7, 131.2, 130.4, 129.6, 126.6, 124.5 (qr, J = 4 Hz), 123.5 (qr, J = 277 Hz) ppm. MS (c.i.): m/z (%) = 259 (0) [M^+ + 18], 242 (100) [M^+ + 1], 222 (78), 197 (50). $\text{C}_{11}\text{H}_6\text{F}_3\text{NO}_2$ (241.17): calcd. C 54.78, H 2.51, N 5.80; found C 54.81, H 2.45, N 5.76.

6-Fluoro-4-trifluoromethyl-3-quinolinecarboxylic Acid (6c): Analogously, from the 2-bromoquinolinecarboxylic acid **5c** (1.7 g, 5.0 mmol); colorless prisms; m.p. $198\text{--}201^\circ\text{C}$ (from acetic acid; decomp.); yield: 0.86 g (66%). ^1H NMR*: δ = 9.11 (s, 1 H), 8.34 (dd, J = 10.2, 5.8 Hz, 1 H), 7.84–7.94 (m, 2 H) ppm. ^{13}C NMR*: δ = 167.6, 162.8 (qr, J = 250 Hz), 148.4 (d, J = 3 Hz), 147.1, 135.7 (qr, J = 32 Hz), 134.5 (d, J = 10 Hz), 124.0, (qr, J = 275 Hz), 123.7 (2 C), 122.3 (d, J = 26 Hz), 109.2 (d, J = 26 Hz, 3 Hz). $\text{C}_{11}\text{H}_5\text{F}_4\text{NO}_2$ (259.16): calcd. C 50.98, H 1.94; found C 51.02, H 1.85. The debromination of the acid **5c** (10 mmol) was also accomplished by transfer hydrogenation using ammonium formate (1.3 g, 20 mmol) as the hydrogen source in methanol (20 mL) in the presence of catalytic amounts of palladium on charcoal at 25°C for 2 h; yield: 1.84 g (71%).

8-Methyl-4-trifluoromethyl-3-quinolinecarboxylic Acid (6g): Analogously, as described for acid **6a**, from the 2-bromo-3-quinolinecarboxylic acid **5g** (1.7 g, 5.0 mmol); colorless prisms; m.p. $174\text{--}176^\circ\text{C}$ (from acetic acid; reprod.); yield: 0.85 g (67%). ^1H NMR*: δ = 9.12 (s, 1 H), 8.09 (d, J = 6.3 Hz, 1 H), 7.84 (d, J = 6.0 Hz, 1 H), 7.77 (t, J = 7.9 Hz, 1 H), 2.82 (s, 3 H) ppm. ^{13}C NMR*: δ = 168.1, 148.9, 147.5, 139.3, 132.1, 130.2, 127.1, 124.3 (qr, J = 276 Hz), 123.2 (qr, J = 4 Hz), 122.8 ppm. MS (c.i.): m/z (%) = 273 (0) [M^+ + 18], 256 (100) [M^+ + 1], 236 (4), 210 (3). $\text{C}_{12}\text{H}_8\text{F}_3\text{NO}_2$ (255.19): calcd. C 56.48, H 3.16, N 5.49; found C 56.25, H 2.94, N 5.46.

Miscellaneous

2-Bromo-7-fluoro-4-trifluoromethyl-8-quinolinecarboxylic Acid (7): When the 2-bromoquinoline **2f** was treated consecutively with lithium diisopropylamide and carbon dioxide as described above (see the conversion of the bromoquinoline **2a** into the acid **5a**), deprotonation and carboxylation occurred at the 8- rather than the 3-position; colorless needles; m.p. $135\text{--}137^\circ\text{C}$ (from ethyl acetate;

decomp.); yield: 4.41 g (87%). ^1H NMR*: δ = 8.35 (tq, J = 7.6, 2.0 Hz, 1 H), 8.14 (s, 1 H), 7.85 (dd, J = 9.6, 8.9 Hz, 1 H) ppm. ^{13}C NMR*: δ = 168.3, 165.0 (d, J = 256 Hz), 147.7, 140.8 (qr, J = 33 Hz), 132.8 (d, J = 10 Hz), 128.7, 128.5 (2 C), 127.3 (qr, J = 274 Hz), 124.5 (d, J = 25 Hz), 123.6 ppm. MS (c.i.): m/z (%) = 356 (0) [M^+ + 18], 339 (21) [M^+ + 1], 338 (22) [M^+], 295 (100), 276 (6), 214 (22). $\text{C}_{11}\text{H}_4\text{F}_4\text{NO}_2$ (338.05): calcd. C 39.08, H 1.19; found C 39.28, H 1.12.

2-Formyl-8-methyl-4-trifluoromethyl-3-quinolinecarboxylic Acid (8):

The 2-bromo-3-quinolinecarboxylic acid **5g** (1.67 g, 5.0 mmol) was treated with butyllithium (10 mmol) in diethyl ether at -100°C as described for the preparation of the debrominated acid **6a**. However, methanol was replaced by *N,N*-dimethylformamide (0.77 mL, 0.73 g, 10 mmol) as the reagent. After 2 h at -75°C , tetrafluoroboric acid–diethyl ether (15 mmol) was added and, after evaporation of the volatiles, the product crystallized from ethyl acetate; colorless prisms; m.p. $217\text{--}219^\circ\text{C}$ (decomp.); yield: 0.81 g (57%). ^1H NMR*: δ = 8.34 (d, J = 8.5 Hz, 1 H), 7.97 (d, J = 7.0 Hz, 1 H), 7.84 (dd, J = 9.1, 7.1 Hz, 1 H), 7.34 (d, J = 7.8 Hz, 1 H), 6.81 (d, J = 6.5 Hz), 2.87 (s, 3 H) ppm. ^{13}C NMR*: δ = 164.0, 152.5, 140.2, 135.6 (qr, J = 35 Hz), 134.9, 133.6, 131.4, 125.2 (qr, J = 5 Hz), 124.9, 124.7 (qr, J = 279 Hz), 98.4, 19.5 ppm. MS (c.i.): m/z (%) = 301 (19) [M^+ + 18], 284 (100) [M^+ + 1], 283 (26) [M^+], 255 (11), 239 (5). $\text{C}_{13}\text{H}_8\text{F}_3\text{NO}_3$ (283.20): calcd. C 55.13, H 2.85, N 4.95; found C 55.11, H 2.91, N 4.93.

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